

Electronic Supporting Information:

Reversible Luminescent Switching Induced by Heat/Water Treatment in a Zero-Dimensional Hybrid Antimony(III) Chloride

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Table S1. Crystal data and structure refinement details for [H₂BPZ][SbCl₅]•H₂O at 293 K.

CCDC number	2220612
Empirical formula	C ₁₁ H ₂₀ Cl ₅ N ₂ OSb
Formula weight	495.29
Temperature/K	293(2) K
Wavelength/Å	0.71073
Crystal system	monoclinic
Space group	<i>P</i> 2 ₁ / <i>c</i>
<i>a</i> /Å	15.1814(14)
<i>b</i> /Å	12.1122(11)
<i>c</i> /Å	10.4217(11)
<i>α</i> /°	90
<i>β</i> /°	103.164(10)
<i>γ</i> /°	90
<i>V</i> /Å ³	1866.0(3)
<i>Z</i>	4
ρ_{calc} g/cm ³	1.763
Absorption coefficient/mm ⁻¹	2.191
<i>F</i> (000)	976
Crystal size/mm ³	0.200 × 0.100 × 0.050
Theta range for data collection /°	2.174 – 29.729
Limiting indices	-20 ≤ <i>h</i> ≤ 20, -16 ≤ <i>k</i> ≤ 16, -12 ≤ <i>l</i> ≤ 12
Reflections collected/ unique	4789/4789
Completeness to theta	25.242, 100.0 %
Absorption correction	Semi-empirical from equivalents
Refinement method	Full-matrix least-squares on <i>F</i> ²
Data/restraints/parameters	4789/2/189
Goodness-of-fit on <i>F</i> ²	1.074
Final <i>R</i> indexes [<i>I</i> ≥ 2σ(<i>I</i>)]	<i>R</i> ¹ = 0.0411, <i>wR</i> ² = 0.0722
Final <i>R</i> indexes [all data]	<i>R</i> ¹ = 0.0785, <i>wR</i> ² = 0.0889
Extinction coefficient	0.00159(18)
Largest diff. peak and hole/e Å ⁻³	0.752, -0.778

[a] $R_1 = \sum \|F_o| - |F_c| \| / \sum |F_o|$, [b] $wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$

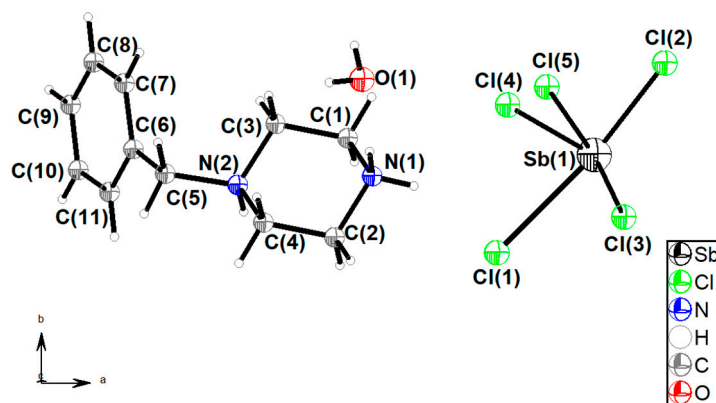


Figure S1. ORTEP drawing (50% ellipsoid probability) of the asymmetric unit of $[\text{H}_2\text{BPZ}][\text{SbCl}_5]\cdot\text{H}_2\text{O}$ at 293 K.

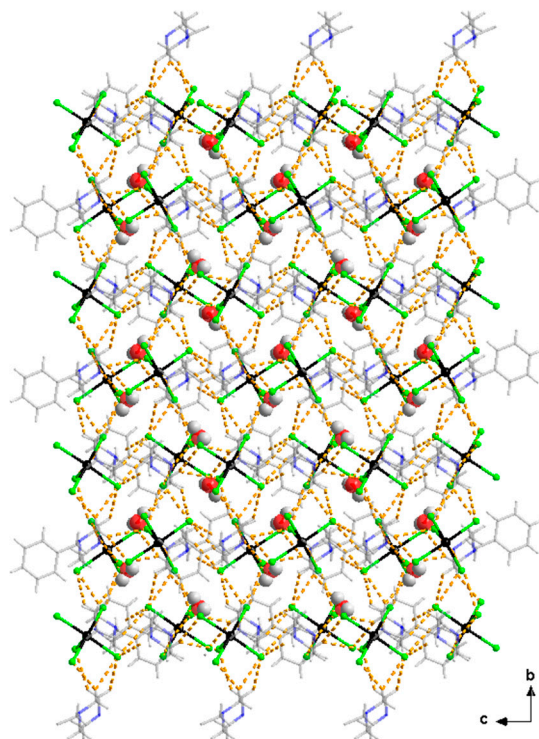


Figure S2. A supramolecular layer in $[\text{H}_2\text{BPZ}][\text{SbCl}_5]\cdot\text{H}_2\text{O}$ viewed along the a axis in which lattice water molecules are located; water molecules are in CPK mode; hydrogen bonds with water molecules are not shown.

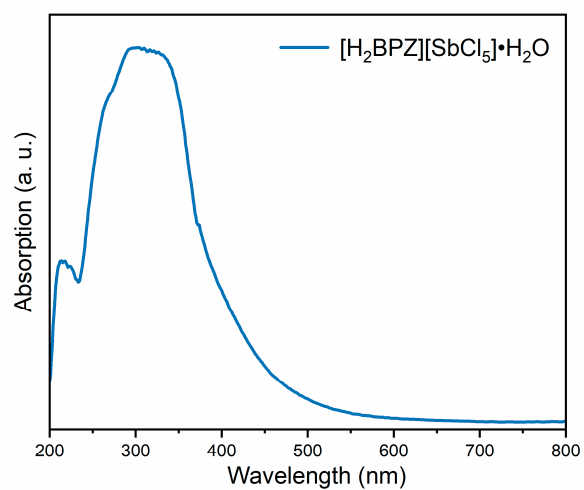


Figure S3. Solid-state UV-visible absorption spectrum of $[\text{H}_2\text{BPZ}][\text{SbCl}_5] \cdot \text{H}_2\text{O}$ at RT.

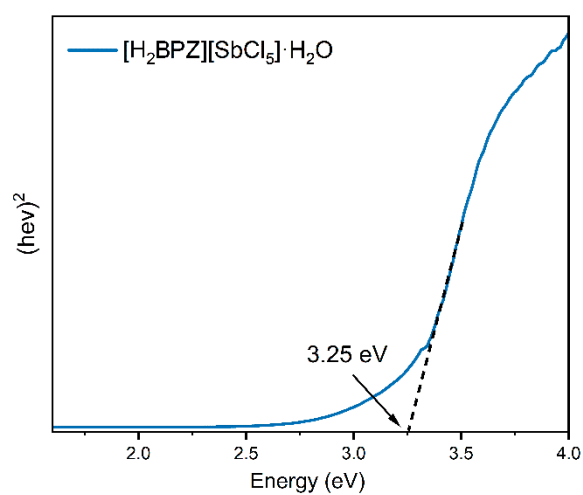


Figure S4. The experimental direct band gap of $[\text{H}_2\text{BPZ}][\text{SbCl}_5] \cdot \text{H}_2\text{O}$ calculated as 3.25 eV.

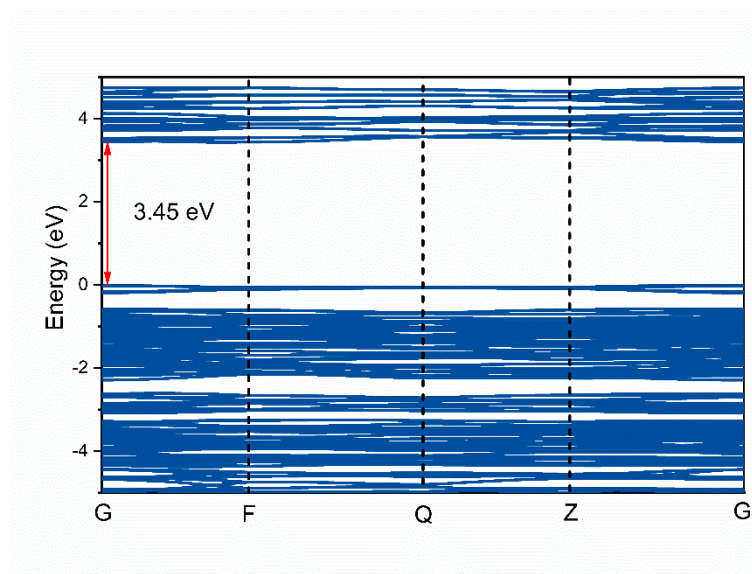


Figure S5. The band structure of $[\text{H}_2\text{BPZ}][\text{SbCl}_5] \cdot \text{H}_2\text{O}$ and the direct band gap is calculated as 3.45 eV.

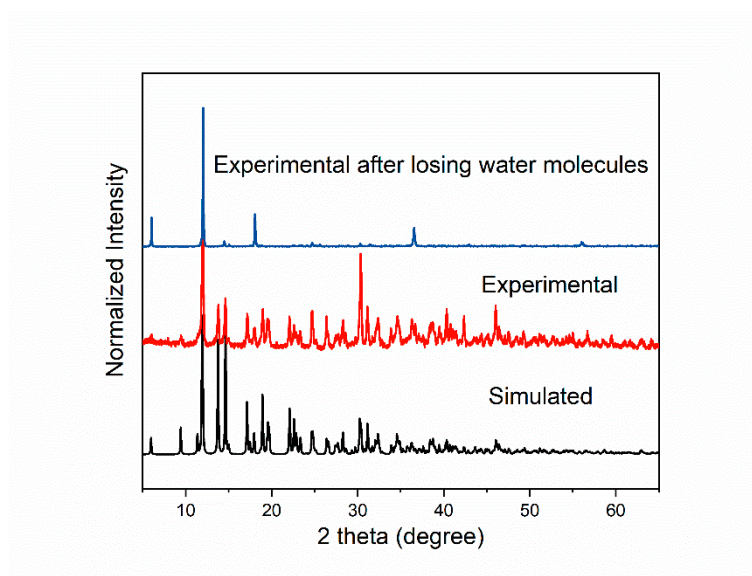


Figure S6. The simulated (bottom) and experimental (middle) PXRD patterns for $[\text{H}_2\text{BPZ}][\text{SbCl}_5] \cdot \text{H}_2\text{O}$, and experimental PXRD pattern for $[\text{H}_2\text{BPZ}][\text{SbCl}_5]$ (top).

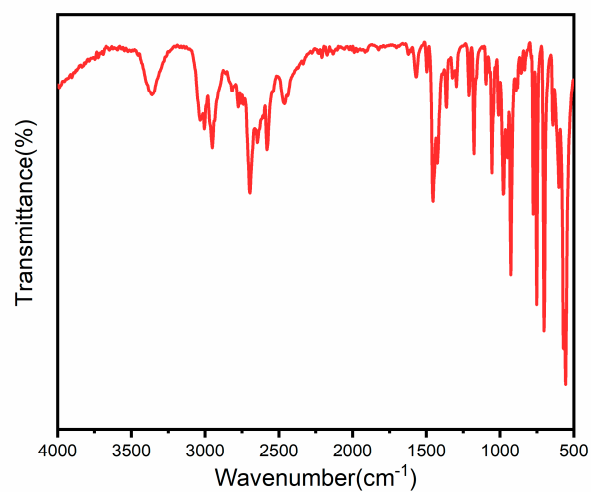


Figure S7. FTIR spectrum of [H₂BPZ][SbCl₅]•H₂O crystal powder at RT.