



Article Abietane-Type Diterpenoids from the Arils of Torreya grandis

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Abstract: A chemical investigation of the arils of *Torreya grandis* led to the isolation of seven abietanetype diterpenoids (compounds 1–7) including three previously undescribed compounds, one unreported natural product, and three known analogs. The structures of these compounds were determined by means of spectroscopy, single-crystal X-ray diffraction, and ECD spectra. An antibacterial activity assay showed that compounds **5** and **6** had significant inhibitory effects on methicillinresistant *Staphylococcus aureus*, with MIC values of 100 μ M. Moreover, compounds **1**, **3**, **4**, and **7** exhibited anti-neuroinflammatory activity in LPS-stimulated BV-2 microglia cells, with the IC₅₀ values ranging from 38.4 to 67.9 μ M.

Keywords: diterpenoids; structure elucidation; antibacterial; anti-neuroinflammatory; Torreya grandis

1. Introduction

Taxaceae plants are the economically and medicinally important coniferous evergreen trees that comprise the well-known genera *Taxus*, *Pseudotaxus* and *Austrotaxus*, whereas the genera *Torreya* and *Amentotaxus* attributed to this family have been controversial for a long time due to their close resemblance to the family of Cephalotaxaceae [1]. Among them, *Torreya* is a primitive member of the gymnospermous yew family (Taxaceae), which consists of seven species and is distributed in the Northern Hemisphere, including North America (*T. taxifolia* and *californica*), Japan (*T. nucifera*), and China (*T. fargesii*, *T. grandis*, *T. jackii*, and *T. yunnanensis*) [2]. However, species of the genus *Torreya* are similar in morphology, and the relationship within this genus is still vague [3].

Torreya grandis Fort. ex Lindl cv. Merrillii Hu (Taxaceae), a native Chinese species naturally distributed across subtropical areas in China, has high nutritional and medicinal value [4,5]. Its leaves and fruits are used in traditional Chinese medicine to cure cough, rheumatism, and malnutrition [4]. Recent chemical investigations of *T. grandis* have led to the discovery of structurally diverse phytochemicals, including diterpenoids [6,7], glycosides [5], flavonoids [8], phenols [9], and fatty acids [10], which exhibit various bioactivities such as antibacterial [6], antioxidant [8,9], and antinociceptive and anti-inflammatory effects [11]; attenuation of cognitive impairment [12]; and inhibitory effects against DGAT1/2 [5]. Nevertheless, chemical constituents of the arils of *T. grandis* have rarely been reported [13,14]. As part of our continuing search for novel and/or bioactive metabolites from Chinese medicinal plants [15–17], a phytochemical investigation of *T. grandis* arils led to the discovery of seven abietane-type diterpenoids (compounds 1–7), of which compounds 1–3 were previously undescribed, compound 4 was an unreported natural product, and compounds 5–7 were known analogs (Figure 1). Herein, the isolation, structural analysis, and bioactivities of these products are reported.



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Figure 1. Chemical structures of compounds 1–7.

2. Results and Discussion

Torregrandin A (1) was obtained as a yellow oil, and its molecular formula was determined as $C_{17}H_{22}O_2$ by a molecular negative-ion peak at m/z 257.1548 [M – H]⁻ (calcd. for C₁₇H₂₁O₂, 257.1547) in its HR-ESI-MS spectrum, requiring 7 degrees of unsaturation. The ¹H NMR data (Table 1) showed characteristic signals of a 1,2,4-trisubstituted benzene ring at $\delta_{\rm H}$ 7.05 (d, J = 8.6 Hz, 1H), 6.56 (dd, J = 8.6, 3.0 Hz, 1H), and 6.43 (d, J = 3.0 Hz, 1H), which were supported by the chemical shifts at $\delta_{\rm C}$ 153.4, 141.5, 136.5, 125.7, 115.2, and 113.3 in the ¹³C NMR spectrum (Table 1). Furthermore, an aldehyde group was indicated by the ¹H NMR signals at $\delta_{\rm H}$ 9.20 (s, 1H), corresponding to the carbon signal at $\delta_{\rm C}$ 206.6 (Table 1). The subsequent interpretation of its 2D NMR spectra (HSQC, COSY, and HMBC) unequivocally underpinned that compound 1 was an abietane-type diterpenoid derivative. Comparing the ¹H and ¹³C NMR data of compound **1** to that of the known 13-hydroxy-8,11,13-podocarpatrien-18-oic acid [18] revealed that the carboxylic acid group located at C-18 was substituted by the aldehyde group in compound 1. This was confirmed by the HMBC correlation of H-18 with C-19 and of H_3 -19 with C-18 (Figure 2). Moreover, the HMBC correlations of H₃-20 with C-1, C-5, and C-9; of H₃-19 with C-5, C-4, and C-3; of H-14 with C-7, C-9, and C-12; of H-12 with C-9, C-13, and C-14; of H-11 with C-10, C-8, and C-13; and of H-7 with C-5, combined with the ${}^{1}H-{}^{1}H$ COSY correlations of H₂-1/H₂-2/H₂-3, of H-5/H₂-6/H₂-7, and of H-11/H-12 unambiguously established the structure of compound **1** (Figure 2). The NOESY correlations of H-3 α ($\delta_{\rm H}$ 1.27) with H-5 ($\delta_{\rm H}$ 1.79), of H-3 β ($\delta_{\rm H}$ 1.40) with H₃-19 ($\delta_{\rm H}$ 1.09), and of H₃-19 ($\delta_{\rm H}$ 1.09) with H₃-20 ($\delta_{\rm H}$ 1.13) established the relative configuration of compound 1, as shown (Figure 3). The similar optical rotations between compound 1 ($[\alpha]^{25}_{D}$ +52.0) and 18-oxoferruginol ($[\alpha]^{25}_{D}$ +69.6) [19] revealed identical configurations of these two compounds. This finding was reinforced by comparing the experimental and calculated ECD spectra of compound 1 (Figure 4), underscoring its (4R,5R,10S)-configuration.



Figure 2. COSY and key HMBC correlations of compounds 1-4.

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	1 ^a		2 ^b	
No.	δ_{C} Type	$\delta_{ m H}$ (J in Hz)	δ _C Type	$\delta_{ m H}$ (J in Hz)
1	38.1	2.24 (m, 1H); 1.77 (m, 1H)	39.5	2.19 (m, 1H); 1.36 (m, 1H)
2	17.8	1.71 (m, 2H)	21.5	1.70 (m, 1H); 1.66 (m, 1H)
3	32.2	1.40 (m, 1H); 1.27 (m, 1H)	43.4	1.80 (m, 1H); 1.43 (m, 1H)
4	49.9		73.2	
5	43.0	1.79 (m, 1H)	53.4	1.55–1.58 (m, 1H)
6	21.4	1.75 (m, 1H); 1.26(m, 1H)	19.3	2.09 (m, 1H); 1.58–1.67 (m, 1H)
7	29.8	2.69 (m, 2H)	30.8	2.78 (m, 2H)
8	136.5		126.8	
9	141.5		148.3	
10	36.2		39.3	
11	125.7	7.05 (d, 8.6, 1H)	111.8	6.63 (s, 1H)
12	113.3	6.56 (dd, 8.6, 3.0, 1H)	153.4	
13	153.4		133.5	
14	115.2	6.43 (d, 3.0, 1H)	127.4	6.75 (s, 1H)
15			27.7	3.17 (m, 1H)
16			23.2	1.16 (d, 7.0, 3H)
17			23.2	1.17 (d, 7.0, 3H)
18	206.6	9.20 (s, 1H)		
19	14.1	1.09 (s, 3H)	22.9	1.19 (s, 3H)
20	25.4	1.13 (s, 3H)	24.9	1.13 (s, 3H)
18-				
OMe				

Table 1. ¹H NMR (400 MHz) and ¹³C NMR (100 MHz) spectroscopic data for compounds 1 and 2.

^a Recorded in CDCl₃. ^b Recorded in CD₃OD. Chemical shifts (δ) are expressed in ppm, and *J* values are expressed in Hz.



Figure 3. NOESY correlations of compounds 1 (left) and 3 (right).



Figure 4. The experimental and calculated ECD spectra of compound 1.

Compound **2** was shown to have a molecular formula of $C_{19}H_{27}O_2$ based on its molecular negative-ion peak at m/z 287.2018 [M – H][–] (calcd. for $C_{19}H_{27}O_2$, 287.2017) in its HR-ESI-MS spectrum. The ¹H and ¹³C NMR data (Table 1) of compound **2** were very similar to those of torregrandol A [20]. Fortunately, compound **2** was crystallized from a methanol solution, which led to the confirmation of its structure and the assignment of its absolute configuration as 4R,5R,10*S*, based on its single-crystal X-ray diffraction analysis (Cu K α) (Figure 5).



Figure 5. X-ray structure of compound 2.

Torregrandin B (compound **3**) shared an identical molecular formula ($C_{19}H_{27}O_2$) with compound **2**, according to its molecular negative-ion peak at m/z 287.2021 [M – H]⁻ (calcd. for $C_{19}H_{27}O_2$, 287.2017) in its HR-ESI-MS spectrum. The ¹H and ¹³C NMR data (Table 2) of this compound were very similar with those of compound **2** except for a downfield signal at δ_C 30.9, ascribable to the 4-CH₃ in the ¹³C NMR spectrum. To the best of our knowledge, the carbon resonances of an α -oriented CH₃ group at C-4 in abietane-type diterpenoids can range from 25 to 35 ppm, whereas in a β -oriented CH₃ group, they can range from 15 to 25 ppm [6,21]. The downfield chemical shift of 4-CH₃ (δ_C 30.9) in comparison with that of compound **2** (δ_C 22.9) revealed that compound **3** was an epimer of compound **2** at C-4. Next, the NOESY correlations of H-3 α (δ_H 1.72) with H-5 (δ_H 1.42) and H₃-19 (δ_H 1.25), of H-5 (δ_H 1.42) and H-1 β (δ_H 2.21), and of H-1 α (δ_H 1.42) with H₃-20 (δ_H 1.30) established the relative configuration of compound **3**, as shown (Figure 3). The absolute configuration was determined by means of ECD calculation, which unambiguously pinpointed the (4*S*,*SR*,10*S*)-configuration of compound **3** (Figure 6).



Figure 6. The experimental and calculated ECD spectra of compound 3.

NT -	3 ^a		4 ^b	
No. —	δ_{C} Type	δ_{H} (J in Hz)	δ_{C} Type	$\delta_{ m H}$ (J in Hz)
1	38.4	2.21 (dt, 13.0, 4.0, 1H); 1.42 (m, 1H)	36.7	2.17 (dd, 12.9, 3.5, 1H); 1.60–1.72 (m, 1H)
2	18.6	1.62 (m, 2H)	17.7	1.60–1.72 (m, 1H); 1.48 (td, 12.9, 3.8, 1H)
3	40.9	1.72 (m, 1H); 1.42 (m, 1H)	36.1	1.76 (t, 12.9, 1H); 1.60–1.72 (m, 1H)
4	72.5		46.2	
5	48.8	1.42 (m, 1H)	44.0	2.45 (dd, 14.1, 3.3, 1H)
6	18.2	2.00 (m, 1H); 1.83 (m, 1H)	37.2	2.68 (dd, 17.6, 14.1, 1H); 1.98 (dd, 17.6, 3.3, 1H)
7	28.8	2.86 (m, 2H)	195.6	
8	127.1		122.3	
9	147.9	2.3 (m, 1H)	155.3	
10	37.3		37.0	
11	110.9	6.64 (s, 1H)	109.3	6.79 (s, 1H)
12	150.9		160.5	
13	131.7		132.9	
14	126.8	6.85 (s, 1H)	125.1	7.64 (s, 1H)
15	27.0	3.12 (m, 1H)	26.1	3.13 (m, 1H)
16	22.7	1.23 (d, 7.0, 3H)	22.2	1.13 (d, 6.4, 3H)
17	22.9	1.24 (d, 7.0, 3H)	22.4	1.14 (d, 6.4, 3H)
18	30.9	1.25 (s, 3H)	177.3	
19			16.2	1.25 (s, 3H)
20	24.5	1.30 (s, 3H)	23.2	1.18 (s, 3H)
18-			52.3	3 59 (c. 3H)
OMe			02.0	3.37 (5, 311)

Table 2. ¹H NMR (400 MHz) and ¹³C NMR (100 MHz) spectroscopic data for compounds 3 and 4.

^a Recorded in CDCl₃. ^b Recorded in DMSO- d_6 . Chemical shifts (δ) are expressed in ppm, and *J* values are expressed in Hz.

Compound **4**, namely methyl 12-hydroxy-7-oxodehydroabietate, was recently isolated from *Torreya grandis* [20], which was previously synthesized by Hamulić and coworkers [22]. However, its absolute configuration was not assigned. In the present work, the absolute configuration of compound **4** was firstly confirmed by the single-crystal X-ray diffraction analysis (Cu K α) (Figure 7).



Figure 7. X-ray structure of compound 4.

The remaining known diterpenoids were characterized as dehydroabietinol (5) [23], dehydroabietic acid (6) [24], and torreyagrandate (7) [25] by comparing the ¹H and ¹³C NMR data as well as mass spectrometric spectra to the published data.

All isolated compounds were tested for their antibacterial activitity against *Escherichia coli*, methicillin-resistant *Staphylococcus aureus* (MRSA), *Pseudomonas aeruginosa*, and *Salmonella*. However, only compounds **5** and **6** had significant inhibitory effects on MRSA, with MIC

values of 100 μ M, which were comparable to those of the positive control, rifampicin (MIC = 0.625 μ M). Other compounds displayed no obvious antibacterial activity at 100 μ M. Moreover, the anti-neuroinflammatory activity in LPS-induced BV-2 cells was evaluated for these compounds. The results indicated that compounds **1**, **3**, **4**, and **7** showed a weak inhibitory effect on NO production, with IC₅₀ values of 49.4, 41.9, 38.4, and 52.6 μ M (Table 3), respectively. Meanwhile, none of the remaining compounds exhibited an inhibitory effect on NO production at 100 μ M. All the results are representative of three independent experiments.

Table 3. Inhibitory effects of compounds **1**, **3**, **4**, and **7** on NO production induced by LPS in BV-2 microglial cells.

Compound	IC ₅₀ (µM) ^a	Cell Viability (%) ^b
1	49.4 ± 0.4	81.6 ± 3.5
3	41.9 ± 1.5	85.4 ± 1.3
4	38.4 ± 0.6	87.1 ± 2.3
7	52.6 ± 2.3	96.1 ± 1.0
quercetin ^c	10.8 ± 1.6	99.7 ± 1.8

^a Results are presented as mean \pm SD from three independent experiments. ^b Cell viability (%) of BV-2 microglial cells at 100 μ M expressed as the mean \pm SD from three independent experiments. ^c Positive control.

3. Materials and Methods

3.1. General

A Rudolph Autopol III instrument was used to measure the optical rotations. HR-ESI-MS spectra were tested using a TripleTOF 5600+ system (AB SCIEX, Framingham, MA, USA). Infrared (IR) spectra were acquired using a Bruker TENSOR 27 spectrometer (Bruker). Ultraviolet (UV) and electronic circular dichroism (ECD) spectra were recorded by a Chirascan spectrometer (Applied Photophysics, Ltd., Leatherhead, UK). The NMR data were acquired on a Bruker Avance-400 spectrometer (Beijing Oubeire Co., Ltd., Beijing, China). Column chromatography (CC) was executed using silica (200–300 mesh, Qingdao Marine Chemical Inc., Qingdao, China), C18 reversed-phase silica (ODS-AQ-HG GEL, AQG12S50, YMC, Co., Ltd., Kyoto, Japan), and Sephadex LH-20 gels (GE Healthcare, Inc., Uppsala, Sweden). An Agilent 1100 series system (Agilent Technologies, Inc., Agilent Technologies, Inc., CA, USA) with a Prazis absolute C18 column (5 μ m, 10 mm × 250 mm) was used for the HPLC analysis and preparation. Fractions were monitored by TLC (Qingdao Marine Chemical, Ltd., Qingdao, China), with the spots visualized using the vanillin–sulfuric acid color method.

3.2. Plant Materials

The arils of *Torreya grandis* Fort. ex Lindl. cv. Merrillii. (Taxaceae) were collected in November of 2022 in Yuexi, Anhui Province, People's Republic of China. The plant sample was identified by Prof. Zhen-Hai Wu at College of Life Sciences, Northwest A&F University. A voucher specimen has been deposited in our institute with the following accession number: TG-arils-2022-AH.

3.3. Extraction and Isolation

The air-dried powder of the arils of *Torreya grandis* (1.5 kg) was presoaked three times with 95% EtOH (3 × 2 L) at room temperature to produce an extract (338 g) which was first re-suspended in H₂O and then partitioned with petroleum ether (PE), ethyl acetate (EtOAc), and *n*-butanol. The EtOAc extract (75 g) was eluted with PE-EtOAc (v/v, 50:1 \rightarrow 1:1), which was then combined under the guidance of the thin-layer chromatography (TLC) analysis to obtain eight fractions (A–H). Fraction E (11.6 g) was subjected to a silica gel.

Gel CC with PE-EtOAc (v/v, 10:1 \rightarrow 2:1) produced five fractions (E1–E5). Fraction E2 (1.2 g) was isolated on a Sephadex LH-20 column eluted with MeOH to yield five fractions (E2a–E2e). Fraction E2b (300 mg) was separated on a reverse-phase C18 silica gel

CC column eluted with MeOH-H₂O (20–100%, v/v) to yield three fractions (E2b1–E2b3). Fraction E2b2 (120 mg) was purified by semipreparative RP-HPLC eluted with CH₃OH– H_2O (60:40, v/v) to yield compounds 2 (7.8 mg, t_R 22.7 min) and 3 (5.4 mg, t_R 23.5 min). Fraction E3 (2.2 g) was separated on a silica gel CC column and eluted with PE-EtOAc (v/v, $10:1 \rightarrow 2:1$) to obtain five fractions (E3a–E3e). Fraction E3c (320 mg) was chromatographed over a Sephadex LH-20 column with MeOH to produce three fractions (E3c1–E3c3). After purification by semipreparative RP-HPLC with CH₃OH–H₂O (60:40, v/v), fraction E3c2 (90 mg) yielded compound 4 (4.8 mg, $t_{\rm R}$ 21.5 min). Fraction E4 (900 mg) was separated by a Sephadex LH-20 column with MeOH to yield three major fractions (E4a-E4c). Fraction E4b (85 mg) was purified by semipreparative RP-HPLC with CH₃OH-H₂O (62:38, v/v) to yield compound 7 (21 mg, $t_{\rm R}$ 28 min). Fraction F (2.5 g) was chromatographed over a silica gel CC column with PE-EtOAc (v/v, 10:1 \rightarrow 1:1) to yield five fractions (F1–F5). Fraction F2 (220 mg) was subjected to a Sephadex LH-20 column with MeOH to yield three fractions (F2a–F2c). Fraction F2b (45 mg) yielded compound 5 (17 mg, t_R 25.7 min) after purification by semipreparative RP-HPLC with CH₃OH-H₂O (60:40, v/v). Fraction G (10.5 g) was subjected to a silica gel CC column with PE-EtOAc (v/v, 10:1 \rightarrow 1:1) to yield five fractions (G1–G4). Fraction G2 (1.8 g) was subjected to a reverse-phase C18 silica gel CC column eluted with MeOH-H₂O (20–90%, v/v) to obtain three fractions (G2a–G2c). After re-purification using a Sephadex LH-20 column, compound 1 (12 mg, $t_{\rm R}$ 19.5 min) was obtained from the fraction G2b (320 mg) through a semipreparative RP-HPLC column with CH₃OH–H₂O (65:35, v/v). Fraction G2c (500 mg) was purified by a Sephadex LH-20 column eluted with MeOH to generate three fractions (G2c1–G2c3). Then, fraction G2c2 (105 mg) was separated by a semipreparative RP-HPLC column with CH₃OH-H₂O (60:40, v/v) to yield compound **6** (30 mg, t_R 31.3 min).

3.4. Structural Elucidation

Torregrandin A (compound 1), yellow oil; $[\alpha]_D^{25}$ +52 (c 0.10, MeOH); UV (MeOH) λ_{max} (log ε) 200 (1.95) nm; IR (KBr) ν_{max} 3427, 3172, 2928, 1717, 1615, 1498, 1399, 1112 cm⁻¹; HR-ESI-MS (negative) m/z 257.1548 [M – H]⁻ (calcd. for C₁₇H₂₁O₂, 257.1547); ¹H and ¹³C NMR dataassigned and listed in Table 1.

Torregrandin B (compound **3**), yellow oil; $[\alpha]_D^{25}$ +59 (c 0.10, MeOH); UV (MeOH) λ_{max} (log ε) 200 (1.73) nm; IR (KBr) ν_{max} 3604, 3304, 2934, 1714, 1507, 1423, 1368, 1236, 1017 cm⁻¹; HR-ESI-MS (negative) m/z 287.2021 [M – H]⁻ (calcd. for C₁₉H₂₇O₂, 287.2017); ¹H and ¹³C NMR data assigned and listed in Table 2.

3.5. Crystal Data for Compounds 2 and 4

Compound **2**, $C_{19}H_{28}O_2$, Mr = 288.41 block from MeOH, space group C_{121} , a = 18.245(3) Å, b = 8.7702 (12) Å, c = 10.5065(15) Å, V = 1591.6 (4) Å³, Z = 4, μ = 0.374 mm⁻¹, and F(000) = 632.0; T = 170.0; crystal dimensions: $0.08 \times 0.06 \times 0.04$ mm³; R = 0.0393, wR = 0.0982, S = 1.090; Flack parameter = 0.17(12); crystallographic data for compound **2** has been deposited at the Cambridge Crystallographic Data Center with the accession number CCDC-2328804.

Compound 4, C₂₁H₂₈O₄, Mr = 344.43 block from MeOH, space group $P_{21}2_{12}$, a = 11.0818(2) Å, b = 11.2194(2) Å, c = 14.5634(2) Å, V = 1810.68(5) Å³, Z = 4, μ = 0.690 mm⁻¹ and F(000) = 744.0; T = 150.0; crystal dimensions: 0.2 × 0.15× 0.1 mm³; R = 0.0713, wR = 0.1545, S = 1.135; Flack parameter = -0.05(5); crystallographic data for compound 4 has been deposited at the Cambridge Crystallographic Data Center with the accession number CCDC-2328805.

3.6. ECD Calculation

The ECD calculation was performed according to our previously reported methods [26].

3.7. Antibacterial Assay

To determine the MICs, bacterial strains of *Escherichia coli*, methicillin-resistant *Staphylococcus aureus* (MRSA), *Pseudomonas aeruginosa*, and *Salmonella* were cultured in 2 mL of Mueller–Hinton broth (MHB) for 3-5 h at 37 °C, followed by a dilution of the mixture to a concentration of 1×10^5 CFU/mL. Next, 100 µL of each dilution and 100 µL of bacterial suspension were successively added to 96-well plates. Subsequently, the tested compounds were added to give the final concentrations of 200, 100, 50, 25, and 12.5 µM, respectively. The positive groups were rifampicin (10, 5, 2.5, 1.25, 0.625 µM), and the negative and blank groups were 200 µL of media and bacterial solution (1×10^5 CFU/mL). After incubation at 37 °C for 16–18 h, the antibacterial effects were observed by the naked eye. The lowest concentration of the tested compounds which completely inhibited the growth of bacteria was defined as the MIC value. The experiment was repeated at least three times.

3.8. Cytotoxicity and Anti-Inflammatory Assays

The bioassays for NO production and cell viability were conducted according to our previously reported methods [27,28]

4. Conclusions

In summary, seven abietane-type diterpenoids (compounds 1–7), including three previously undescribed ones and one unreported natural product as well as three known analogs, were isolated and characterized from the arils of *T. grandis*. Moreover, compounds **5** and **6** showed mild inhibitory effects against MRSA, and compounds **1**, **3**, **4**, and **7** exhibited weak anti-neuroinflammatory activity in LPS-induced BV-2 microglia cells. These findings not only enrich the molecular diversity of abietane-type diterpenoids but also offer evidence for antibacterial and anti-neuroinflammatory agents that could be used against human pathogenic bacteria and neuroinflammation-related diseases.

Supplementary Materials: The following supporting information can be downloaded at: https:// www.mdpi.com/article/10.3390/molecules29091905/s1, Figures S1-S7: The 1D and 2D NMR spectra of compound 1 in CDCl₃, Figure S8: The HR-ESI-MS spectrum of compound 1, Figures S9 and S10: The IR and UV spectra of compound 1, Figures S11–S16: The 1D and 2D NMR spectra of compound 2 in CD₃OD, Figure S17: The HR-ESI-MS spectrum of compound 2, Figures S18 and S19: The IR and UV spectra of compound 2, Figures S20–S26: The 1D and 2D NMR spectra of compound 3 in CDCl₃, Figure S27: The HR-ESI-MS spectrum of compound 3, Figures S28 and S29: The IR and UV spectra of compound 3, Figures S30–S32: The 1D NMR spectra of compound 4 in DMSO-d₆, Figure S33: The HR-ESI-MS spectrum of compound 4, Figures S34 and S35: The IR and UV spectra of compound 4, Figures S36 and S37: The 1D NMR spectra of compound 5 in CDCl₃, Figures S38 and S39: The 1D NMR spectra of compound 6 in CDCl₃, Figures S40 and S41: The 1D NMR spectra of compound 7 in CDCl₃. Table S1: Conformational analysis of the B3LYP/6-31G(d) optimized conformers of 1 in the gas phase, Tables S2 and S3: Atomic coordinates (A) of 2 conformers of 1 obtained at the B3LYP/6-31G(d) level of theory in the gas phase, Tables S4 and S5: Key transitions, oscillator strengths, and rotatory strengths in the ECD spectra of 2 conformers of 1 at the CAM-B3LYP/6-311G(d) level of theory in MeOH with IEFPCM solvent model, Table S6: Conformational analysis of the B3LYP/6-31G(d) optimized conformers of 3 in the gas phase, Tables S7–S13: Atomic coordinates (Å) of 7 conformers of 3 obtained at the B3LYP/6-31G(d) level of theory in the gas phase, Tables S14–S20: Key transitions, oscillator strengths, and rotatory strengths in the ECD spectra of 7 conformers of 3 at the CAM-B3LYP/6-311G(d) level of theory in MeOH with IEFPCM solvent model.

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